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In chemistry, a mole is a unit of measurement, such that one mole of a substance contains 6.022×10^{23} particles. In chemistry, particles can be atoms, molecules, or compounds.

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Mole-to-Mole Ratios and Calculations of a Chemical Equation

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representative particles. $49 \times 10^{-3} \text{ mol/L}$
What is the value of K_c ? 2. Chemistry is the study of matter and the changes it undergoes.

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research in condensed matter and materials sciences. Areas of interest broadly include experimental and theoretical research on electronic, magnetic, spectroscopic and structural properties as well as the statistical mechanics and thermodynamics of materials.

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S.No. Units. Topics. 1. Classification of Elements and Periodicity in Properties. Modern periodic law and present form of the periodic table, s, p, d and f block elements, periodic trends in properties of elements atomic and ionic radii, ionization enthalpy, electron gain

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enthalpy, valence, oxidation states and chemical reactivity.

JEE Mains Chemistry Syllabus 2021 with Weightage

A molecule is a group of two or more atoms held together by attractive forces known as chemical bonds; depending on context, the term may or may not

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include ions which satisfy this criterion. In quantum physics, organic chemistry, and biochemistry, the distinction from ions is dropped and molecule is often used when referring to polyatomic ions.. In the kinetic theory of gases, the term ...

Molecule - Wikipedia

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