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Colligative Properties.

Section 1- Compounds

in Aqueous Solutions

dissociation is the

separation of ions that

occurs when an ionic

compound dissolves

see sample problem A

on page 436 do

practice problem #1 on

page 436 Solubility

precipitation reactions

occur when

combinations of ions in

a solution have an

extremely low

solubility and a

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precipitate forms see  
table 1 on page 437 of  
the textbook for  
general solubility  
guidelines Net ...

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dissociation. net ionic  
equation. spectator  
ions. ionization. the  
separating of a molecule  
into simpler molecules,

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atoms, rad.... an  
equation that includes  
only those compounds  
and ions that u.... ions  
that are present in a  
solution in which a  
reaction is tak....  
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### SECTION 13 SHORT

ANSWER 1 c 2 a 3 b 2 a

alcohol b water c the  
gels 3 The mixture is a  
colloid The properties  
are consistent with  
those reported in

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Section 1. dissociation.

net ion equation.

spectator ions.

ionization. the

separation of ions that

occurs when an ionic

compound diss....

includes only those

compounds and ions

that undergo a

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chemical.... ions that  
do not take part in a  
chemical reaction and  
are found....

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Chemistry Is a Physical  
Science What is  
chemistry?

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ionic compound is NOT

represented by a

molecular formula

because an ionic

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compound: ... An ionic  
compound is NOT  
represented by a  
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because an ionic  
compound: Holt  
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### **Quiz - Quizizz**

Modern Chemistry 6

The Periodic Law

CHAPTER 5 REVIEW

The Periodic Law

SECTION 3 SHORT

ANSWER Answer the

following questions in  
the space provided. 1.

\_\_\_\_\_ When an electron  
is added to a neutral  
atom, energy is (a)  
always absorbed.

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**Chapter 5 Review**

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Matter

## Section 3 Review

### **CH105: Chapter 1 - Measurements in**

### **Chemistry - Chemistry**

#### CHAPTER 6 REVIEW

#### Chemical Bonding

#### SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule.

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(d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b ...

## **6 Chemical Bonding**

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CHAPTER 10 REVIEW

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States of Matter

SECTION 4 SHORT

ANSWER Answer the following questions in the space provided.

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2 REVIEW

Measurements and  
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Answer the following questions in the space provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model.

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concept number 10.

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alCulations and

equations 10, that can

be made from  $109 \times$

104 kilograms of

phosphorus in the

second of these three

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reactions The answer is

$250 \times 10^4 \text{ kg P}$  40 10

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